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# Note

# High-performance liquid chromatographic system for the separation of basic dyes

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Comparatively few papers are devoted to the high-performance liquid chromatographic (HPLC) analysis of basic<sup>1</sup> compared to acid dyes<sup>2,3</sup> and most of this literature is concerned with individual or a narrow range of basic dyes<sup>1,4</sup>.

The aim of this study was to develop an HPLC system which would be capable of separating and detecting a range of synthetic basic dyes used in the dyeing of acrylic fibres.

A silica column with a buffered (pH 9.7) water-methanol eluent was found to provide an acceptable separation system for the twenty-one commercial basic dyes available. A gradient elution program was used which varied the buffer concentration while maintaining a 9:1 ratio of methanol and buffer.

# EXPERIMENTAL

The equipment consisted of two Waters Assoc. Model 510 HPLC pumps, a Water Assoc. Model 680 gradient controller and a Shandon (160  $\times$  4.5 mm I.D.) stainless-steel column. This was packed with Phase Sep Spherisorb silica (5  $\mu$ m). A Rheodyne valve injector fitted with a 20- $\mu$ l sample loop was used. Detection was mainly by a Waters 990 photodiode array detector coupled to a NEC-APC111 computer and Waters 990 printer/plotter. Initial experiments were carried out using a Pye-Unicam LC–UV detector and Phillips PM8251 single pen recorder.

UV-VIS spectra were determined between 200 and 800 nm on a Hewlett-Packard HP 8450 diode array spectrophotometer using a 1-cm pathlength.

# Reagents

Mobile phase. A volume of 94 ml of concentrated ammonia and 21.6 ml of concentrated acetic acid were added to 884 ml of glass distilled water and the pH adjusted to pH 9.76 using either conc. ammonia or conc. acetic acid. The buffer was diluted with glass distilled water (1:3) as required for solvent B. 100 ml of buffer was mixed with 900 ml of HPLC-grade methanol (Fisons) and the eluents filtered under vacuum through a Gelman 0.2- $\mu$ m FF Vericel <sup>TM</sup> membrane filter. All solvents were degassed for 30 min in the Decon FS200 ultrasonic bath before use.

*Standards.* Fifteen basic dyes were obtained from Ciba Geigy and nine from Bayer U.K. (see Table I). Basic blue 7 was obtained from Aldrich. Solutions of these

Basic dye	Manufacturer	RRT*	$\lambda_{max}$ (methanol) (nm) 642 (596)	
Blue 3	Ciba Geigy/Bayer	0.340		
Blue 7	Aldrich	0.112	619	
Blue 124	Ciba Geigy	0.335	612	
Blue 151	Ciba Geigy	0.284	618	
Blue FGRL	Bayer	0.330 (0.225)**	600	
Blue GRL	Ciba Geigy	0.266 (0.298)**	616	
Yellow 21	Bayer	0.261	416	
Yellow 28	Ciba Geigy/Bayer	0.390	444	
Yellow 40	Ciba Geigy	0.352	434	
Yellow 87	Ciba Geigy	0.710	414	
Yellow 91	Ciba Geigy	0.705	434	
Red 14	Bayer	0.290	522	
Red 46	Bayer	0.308 (0.587)**	526	
Red B-LN	Ciba Geigy	0.304 (0.570)**	534	
Red 18:1	Ciba Geigy/Bayer	1.01 (0.588)*	466	
Red 27	Ciba Geigy	0.265 (0.194, 0.286)**	534	
Red 51	Ciba Geigy	0.465 (0.263)**	524	
Red 109	Ciba Geigy	0.388 (0.286)**	514	
Red GRL	Ciba Geigy	0.278 (0.557)**	526	
Red 46	Ciba Geigy	0.364 (0.313, 0.232)**	522	
Violet 16	Bayer	0.182 (0.286)**	552	

### TABLE I

# **RRT OF THE BASIC DYES WITH RESPECT TO BLUE 45**

\* Slight variation of RRT occurred when the column was repacked.

\*\* Minor peak.

dyes were prepared by dissolving 100 mg of dye in 100 ml of HPLC-grade methanol and further diluting to the required concentration, *e.g.* 100  $\mu$ l to 10 ml gave 100 ng in 10  $\mu$ l injection.

## Chromatographic conditions

Solvent A: 90% methanol, 10% ammonium acetate solution-water (1:3); solvent B: 90% methanol, 10% ammonium acetate solution pH 9.76. For gradient programme, see Table II.

# TABLE II

## GRADIENT PROGRAMME

Time (min)	Flow (ml/min)	% A	% B	Curve*
Initial	1.00	100	0	<u></u>
20.00	2.00	50	50	6
26.00	2.00	31	69	6
35.00	1.00	100	0	6

\* Curve 6 denotes a linear gradient.

# **RESULTS AND DISCUSSION**

Initial studies were carried out using the single-wavelength detector at 280 nm. This wavelength was determined from the UV–VIS spectra as suitable for the detection of all the components of the dye mixture. The relative retention times (RRT) of the dyes were calculated using basic blue 45 as the internal standard and the results are shown in Table I and Fig. 1. Seventeen peaks were obtained when the mixture of twenty-one dyes was injected onto the HPLC system, indicating that some of the dyes were not completely resolved by this system. For example, the peak at 17 min contains two components basic yellow 87 and basic yellow 91 (Fig. 2).

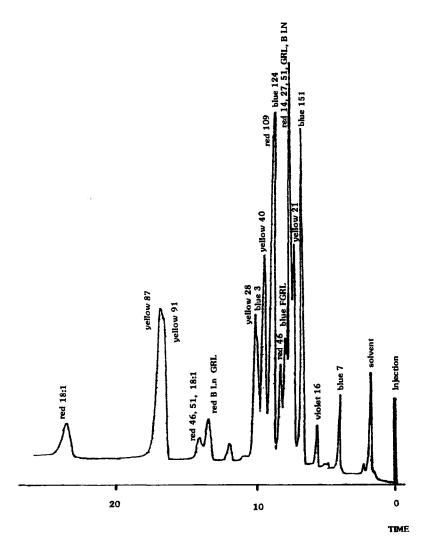


Fig. 1. Chromatogram illustrating the separation of twenty-one basic dyes.

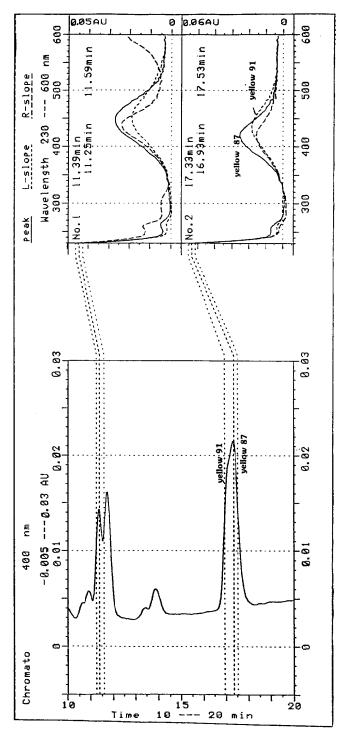


Fig. 2. Diagram showing the identification of dyes by UV-VIS spectroscopic analysis of unresolved peaks.

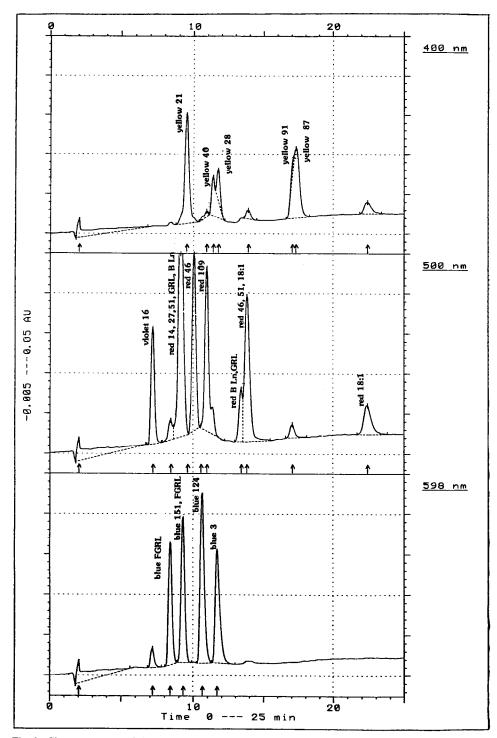


Fig. 3. Chromatograms of the dye mixture at three detecting wavelengths.

To overcome this overlap problem and provide additional information the dyes were re-analysed using the diode array detector. The chromatograms derived from three wavelengths, 400, 500 and 598 nm (yellow, red and blue dyes respectively),

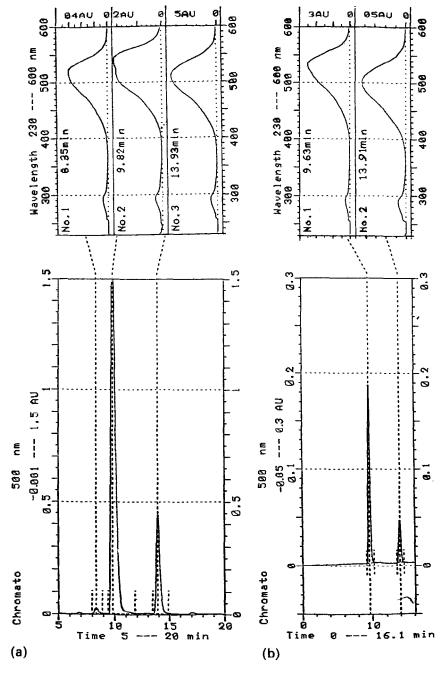


Fig. 4. Chromatograms showing components of (a) basic red B-Ln and (b) basic red GRL.

are shown in Fig. 3. It can be seen that four yellow peaks, four blue peaks and six red peaks were obtained for the dye mixture. Analysis of the UV-VIS spectra across individual peaks indicated those with more than one component. In the case of the

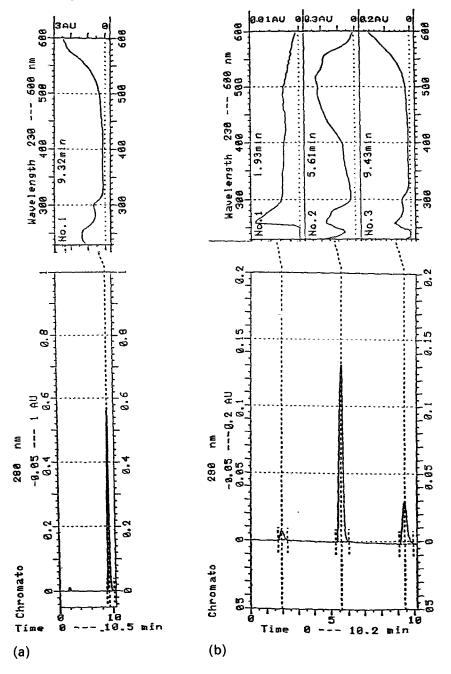


Fig. 5. Chromatograms of basic blue 151 (a) freshly prepared solution and (b) one week old solution.

yellow dyes, the peak at 17 min was shown to contain 2 dyes with  $\lambda_{max}$  of 420 nm (basic yellow 87) and 435 nm (basic yellow 91) (Fig. 2). Similar analysis of the chromatograms at 500 and 598 nm resulted in a total of eighteen out of twenty-one dyes being identified.

Impurities and secondary peaks were detectable by this system, *e.g.* basic red B-Ln and basic red GRL both have two distinct components but the two dyes are not resolved when co-injected. The difference in the UV-VIS spectra for the early eluting components are shown in Fig. 4.

Basic blue 151 was found to decompose in methanol solution to give a red solution over a period of about one week. This decomposition could be monitored by the HPLC system (Fig. 5). These experiments demonstrate the applicability of this HPLC system to a range of problems.

### CONCLUSION

The combined use of the diode array detector with the chromatographic conditions described above provides a powerful system for the separation and detection of basic dyes used in the dyeing of acrylic fibres.

## ACKNOWLEDGEMENTS

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